Synthesis via a Bicyclic β-Lactam of (αS,2R/αR,2S)-α-Aminopiperidine-2-acetic Acid, an Amino-acid Related to the Anti-tumour Agent '593A' [3,6-Bis-(5-chloro-2-piperidyl)piperazine-2,5-dione]

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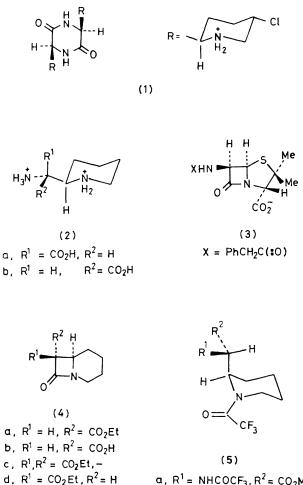
Summary A stereoselective synthesis of $(\alpha S, 2R/\alpha R, 2S)$ - α -aminopiperidine-2-acetic acid (2a) from ethyl 8-oxo-lazabicyclo[4 2 0]octane-7-carboxylate (4a) is described

We have undertaken the synthesis of the bifunctional alkylating agent 3,6-bis-(5-chloro-2-piperidyl)piperazine-2,5-dione ['593A' or cyclo-streptolutyl-streptolutyl, (1)] produced by Streptomyces griseoluteus ¹ This compound is a potential anti-cancer drug² and its stereoisomers, homologues, and radiolabelled derivatives are required for biological evaluation. We now report a straightforward synthesis of both diastereoisomers of α -aminopiperidine-2-acetic acid [(2a) and (2b)], the parent amino-acid of (1), thus paving the way to (1) and many analogues.

There is a correlation between the structure and stereochemistry of (1) and the penicillins, eg penicillin G (3) Strategies for the synthesis of β -lactam antibiotics and their nuclear analogues should therefore be adaptable to the preparation of (1) and analogues We have sought to extend Lowe's method⁵ for preparing nuclear analogues of penicillins to the synthesis of structures related to (1)

Racemic ethyl 8-oxo-l-azabicyclo[4 2 0]octane-7-carboxylate⁵ (4a) was hydrolysed (1 equiv KOH in aq dioxan left overnight at room temperature) to the acid (4b) \ddagger Several attempts, using literature procedures,^{5,6} to replace the carboxy-group of (4b) by an amino group failed Treating (4a) with NaH (1 equiv) in dimethyl sulphoxide easily generated the anion (4c), which was characterised by its ¹H nmr spectrum and by kinetic protonation (rapid addition to excess of dil acetic acid) to (4a) + (4d) (9:1) Reaction of the anion (4c) [generated by treating (4a) with lithium di-isopropylamide in tetrahydrofuran] with toluene-p-sulphonyl azide followed by chlorotrimethylsilane (procedure of ref 7) gave the azides (4e); and (4f): [the mixture was purified by 'flash chromatography'⁸ on silica gel 60 (elution with ether)] Reduction (excess of $\rm H_2S$ and triethylamine' in $\rm CH_2Cl_2$ for 15 min at room temperature) of (4e) + (4f) gave the amino-esters (4g) ‡ and (4h) ‡ purified by short column chromatography on silica gel (gradient elution with ether-methanol) Saponification (1.1 equiv NaOH in ethanol left overnight at room temperature) of (4g) + (4h) gave $(4i)^{+}_{+}$ which separated directly from the reaction mixture in 78% yield Heating (4i) in 5 м hydrochloric acid (5 h, 110 °C) gave a 2:1 mixture of the diastereoisomers [54% overall yield from (4a)], characterised as (2a) and (2b), respectively, by comparing their ¹H n m r spectra in 2 M DCl/D₂O with spectra¹⁰ for threo- and erythro-2,3-diaminobutanoic acid, respectively The predominant diastereoisomer (2a) shows a narrow

doublet $(J \ 3 \ \text{Hz})$ for α -H, as expected^{10,11} for the diprotonated structure (**2a**) in the preferred conformation shown (protonated amino groups *anti*) (*cf* J **3**·6 Hz for α -H of *threo*-2,3-diaminobutanoic acid under similar conditions)¹⁰ The (α S,2R)-isomer of (**2a**) has the same chirality as at C-3 and C-2 of (**1**) The minor diastereoisomer (**2b**) shows a relatively wide doublet (J 7 Hz) for α -H which is consistent with the diprotonated structure (**2b**) existing in the preferred conformation shown (*cf* J 6 6 Hz for α -H of *erythro*-2,3-diaminobutanoic acid)¹⁰



a, $R^1 = NHCOCF_3$, $R^2 = CO_2Me$ b, $R^1 = CO_2Me$, $R^2 = NHCOCF_3$

† Since submission of this manuscript, the synthesis of racemic (1) has been reported (T Fukuyama, R K Frank, and C F Jewell, J Am Chem Soc, 1980, 102, 2122) via a monocyclic β -lactam

 $e, f, R^1, R^2 = CO_2 Et, N_3$

 $g,h,R^1,R^2 = CO_2Et, NH_2$

 $I, R^{1}, R^{2} = CO_{2}^{-}Na^{+}, NH_{2}$

[‡] New compounds had spectroscopic characteristics in accord with their assigned structure.

The mixture of amino-acids (2a) and (2b) was converted into their bis-N-trifluoroacetyl methyl esters (5a); and (5b), respectively, which were easily separated [60%] recovery based on (2a) and (2b)] by 'flash chromatography' on silica gel 60 (elution with ether-petrol, 1:1). The predominant diastereoisomer shows Rf 0.27 (t.l.c. Kieselgel GF₂₅₄, elution with ether-petrol, 1:1); m.p. 122.5-123.5 °C; δ (CDCl₃) 1·4-1·9 (m, CH₂CH₂CH₂), 3·31 (sextuplet, H-6_{ax}), 3.83 (m, H-6_{eq}), 3.83 (s, OCH₃), 4.88 (d of t, J 11, 3, and 3 Hz, $H-2_{eq}$), 5.15 (dd, J 10 and 11 Hz, α -H), and 7.13 (d, J 10 Hz, NH). As expected¹² for a 2-substituted Nacylpiperidine the preferred conformation of (5a) is a chair

Conversion of (5a) into de-chloro-(1) and extension of the described route to (1) and a variety of analogues is in progress.

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¹ B. H. Arison and J. L. Beck, *Tetrahedron*, 1973, 29, 2743; G. R. Pettit, R. B. Von Dreele, D. L. Herald, M. T. Edgar, and H. B. Wood, J. Am. Chem. Soc., 1976, 98, 6742.

- ² J. A. Gottlieb, E. J. Freireich, G. P. Bodey, V. Rodriguez, K. B, McCredie, and J. U. Gutterman, Proc. Am. Assoc. Cancer Res., 1975, 16, 86; S. E. Jones, W. G. Tucker, A. Haut, B. L. Tranum, C. Vaughn, E. M. Chase, and B. G. M. Durie, Cancer Treatment Reports, 1977, 61, 1617.

 - ⁶ G. R. Krow and C. Johnson, Synthesis, 1979, 50.
 ⁴ M. T. Edgar, G. R. Pettit, and T. S. Krupa, J. Org. Chem., 1979, 44, 396.
 ⁵ G. Lowe and J. Parker, Chem. Commun., 1971, 577; D. M. Brunwin, G. Lowe, and J. Parker, J. Chem. Soc. (C), 1971, 3756.
 - ⁶ K. Ninomiya, T. Shioiri, and S. Yamada, Chem. Pharm. Bull., 1974, 22, 1398.
 - ⁷ K. Kuhlein and H. Jensen, Justus Liebigs Ann. Chem., 1974, 369.
- ⁸ W. Clark Still, M. Kahn, and A. Mitra, J. Org. Chem., 1978, 43, 2923.
 ⁹ cf T Adachi, Y. Yamada, I. Inoue, and M. Saneyoshi, Synthesis, 1977, 45.
 ¹⁰ E. Atherton and J. Meienhofer, Z. Physiol. Chem., 1973, 354, 689; W. K. Hausmann, D. B. Borders, and J. E. Lancaster, J. Antibiol., Ser. A, 1969, 22, 207.
- ¹¹ Concerning the application of the Karplus relationship to amino-acids see R. B. Martin, J. Phys. Chem., 1979, 83, 2404.
- ¹² D. J. Hart, J. Am. Chem. Soc., 1980, 102, 397, and references cited therein.